

A synthetic route to a novel type of conformationally constrained *N*-aryloxazolidinones

Rosa Griera,^a Carme Cantos-Llopart,^a Mercedes Amat,^a Joan Bosch,^{a,*}
Juan-C. del Castillo^b and Joan Huguet^b

^aLaboratory of Organic Chemistry, Faculty of Pharmacy, University of Barcelona, 0802-Barcelona, Spain

^bLaboratorios Lesvi S.L., 08970-Sant Joan Despí, Barcelona, Spain

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Abstract—The synthesis of *N*-aryloxazolidinone **1**, a conformationally constrained analog of linezolid embodying a tricyclic pyrrolo[1,2-*a*][4,1]benzoxazepine moiety as the *N*-aryl substituent, is reported. The synthetic route involves the successive construction of the pyrrole, oxazepine, and oxazolidinone rings, with incorporation of the isoxazolylamino moiety in the last synthetic steps. © 2005 Elsevier Ltd. All rights reserved.

The emergence of bacterial resistance to antibiotics is a problem of ever increasing significance.¹ *N*-aryl oxazolidinones have recently attracted a great deal of attention as a new class of orally active, synthetic antibacterial agents with activity against Gram-positive and anaerobic bacteria.² DuP 721 was the first drug candidate of this family of totally synthetic antibiotics,³ whereas linezolid was the first member of this series introduced in the market (Fig. 1).⁴ The novel mechanism of action of linezolid, involving the inhibition of bacterial protein synthesis at a very early stage,⁵ combined with its biological activity against resistant organisms has stimulated further synthetic work in the area.⁶ Among the many different approaches to modify the chemical structure of linezolid, two different strategies have received particular attention: the synthesis of conformationally constrained analogs⁷ and the modification

of the acetamidomethyl group present at C-4 of the oxazolidinone ring.⁸

We present here the synthesis of *N*-aryl oxazolidinone **1** as the first example of a new class of conformationally constrained analogs of linezolid, bearing a tricyclic pyrrolo[1,2-*a*][4,1]benzoxazepine moiety as the aryl substituent on the nitrogen of the core oxazolidinone ring. Additionally, **1** incorporates an *N*-(isoxazol-3-yl)amino-methyl substituent at the C-4 position of the oxazolidinone ring instead of the acetamidomethyl group present in linezolid. The synthetic route is depicted in Scheme 1 and involves, as the key steps, the successive construction of the pyrrole, oxazepine, and C-4 substituted oxazolidinone rings, with incorporation of the isoxazolylamino moiety in the last synthetic steps from the key intermediate **12**.

The pyrrole ring was constructed via Clauson–Kaas reaction⁹ by refluxing a solution of methyl 2-amino-5-nitrobenzoate (**2**)¹⁰ and 2,5-dimethoxytetrahydrofuran in glacial acetic acid. *N*-arylpyrrole **3** was obtained in 82% yield.

The closure of the oxazepine ring was accomplished by intramolecular dehydration of a diol, as previously reported for the construction of the basic pyrrolo[1,2-*a*][4,1]benzoxazepine ring system.¹¹ Thus, formylation of **3** under Vilsmeier–Haack conditions occurred regioselectively on the 2-position of the pyrrole ring to give 2-formylpyrrole **4** (65%) as the major product. Then,

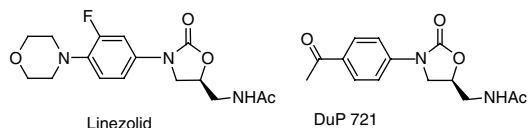
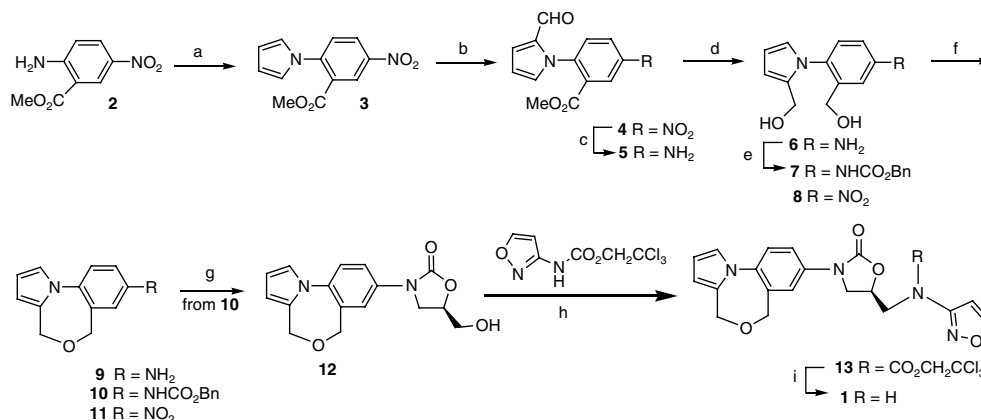


Figure 1.

Keywords: *N*-Aryloxazolidinones; Antibiotics; Linezolid; Constrained analogs; Pyrrolo[1,2-*a*][4,1]benzoxazepine.

* Corresponding author. Tel.: +34 93 402 45 38; fax: +34 93 402 45 39; e-mail: joanbosch@ub.edu



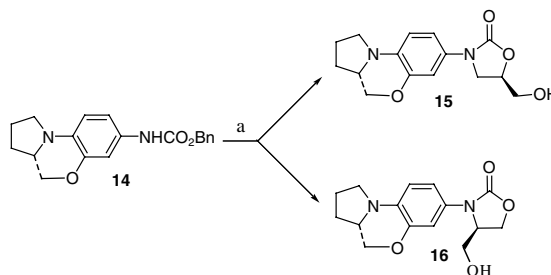
Scheme 1. Reagents and conditions: (a) 2,5-dimethoxytetrahydrofuran, AcOH, reflux; (b) POCl₃, DMF, reflux; (c) H₂, PtO₂, MeOH, rt; (d) LiAlH₄, THF, rt; (e) ClCO₂Bn, Na₂CO₃, acetone, 0 °C to rt; (f) P₂O₅, toluene, 60 °C; (g) (*R*)-glycidyl butyrate, *n*-BuLi, THF, –78 °C to rt, then MeOH, K₂CO₃, rt; (h) PPh₃, DIAD, THF, rt; (i) Zn, AcOH, THF/H₂O, rt.

the nitro group of **4** was chemoselectively reduced by catalytic hydrogenation over PtO₂ to give aniline **5** in 90% yield. A subsequent LiAlH₄ reduction of **5** brought about the reduction of both the formyl and ester groups to give the required aminodiol **6** in 80% yield. Although direct P₂O₅-promoted cyclodehydration of **6** gave tricyclic aniline **9** in poor yield (17%), the closure of the oxazepine ring was satisfactorily accomplished from carbamate **7**, which was easily accessible by treatment of **6** with benzyl chloroformate. The tricyclic carbamate **10** was obtained in 60% overall yield from **6**. An alternative sequence involving the initial reduction of **4** with either LiAlH₄ or Red-Al, followed by cyclodehydration of the resulting diol **8** with P₂O₅ to give **11** took place in poor yields.

For the construction of the 5-substituted oxazolidinone ring we took advantage of the carbamate function present in **10** and used the previously reported regioselective lithium-ion dependent alkylation–cyclization of aryl *N*-lithiocarbamates with (*R*)-glycidyl butyrate.^{4,12–14} Thus, the tricyclic aryl carbamate **10** was treated with *n*-BuLi, and the resulting lithiated intermediate was allowed to react with (*R*)-glycidyl butyrate to furnish, after in situ hydrolysis of the ester function, 5(*R*)-(hydroxymethyl)oxazolidinone **12**¹⁵ in 51% overall yield. Variable amounts (<10%) of the corresponding regioisomeric 4-substituted oxazolidinone were also formed. The formation of mixtures of 4- and 5-(hydroxymethyl)oxazolidinones from (*R*)-glycidyl butyrate have been reported¹² when using *N*-potassium or sodium (but not lithium) carbamate salts. An initial nucleophilic attack of the carbamate anion on the substituted C-1, rather than C-2, position of the oxirane ring accounts for the formation of the unexpected 4-substituted isomer.

It is worth mentioning that we also isolated a similar C-4 substituted oxazolidinone **16** as a by-product (5% yield), in addition to the previously reported 5-substituted isomer **15**^{7c} (45%), from the reaction of tricyclic carbamate **14** with (*R*)-glycidyl butyrate (Scheme 2).¹⁶

The 4- and 5-substituted regioisomeric oxazolidinones were easily distinguished by ¹H and ¹³C NMR from



Scheme 2. Reagents and conditions: (a) (*R*)-glycidyl butyrate, *n*-BuLi, THF, –78 °C to rt, then MeOH, K₂CO₃, rt.

the chemical shift of the oxazolidinone methine and methylene groups.

Finally, the 3-aminoisoxazole moiety was incorporated by Mitsunobu reaction of alcohol **12** with 3-(2,2,2-trichloroethoxycarbonylamino)isoxazole, followed by deprotection of the resulting trichloroethyl carbamate **13** with zinc in acetic acid. The target oxazolidinone **1**¹⁷ was obtained in 52% overall yield from **12**.

The efficient route reported above for the preparation of **1** can provide facile and practical access to structurally diverse new tricyclic oxazolidinones, since the hydroxymethyl substituent of the key intermediate **12** can be further elaborated into a variety of methylene *O* and *N* linked substituents at the C-5 position of the oxazolidinone ring.⁸

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15. Spectroscopic data for **12**: ^1H NMR (300 MHz, CD_3OD) δ 3.88 (dd, $J = 12.4$, 4.0 Hz, 1H), 4.03 (dd, $J = 12.4$, 3.4 Hz, 1H), 4.14 (dd, $J = 6.4$, 2.6 Hz, 1H), 4.32 (app t, $J = 8.9$ Hz, 1H), 4.56 (s, 2H), 4.58 (s, 2H), 4.94–5.03 (m, 1H), 6.42–6.46 (m, 2H), 7.27–7.29 (m, 1H), 7.63 (d, $J = 8.6$ Hz, 1H), 7.84 (d, $J = 2.6$ Hz, 1H), 7.90 (dd, $J = 8.6$, 2.6 Hz, 1H); ^{13}C NMR (75 MHz, CD_3OD) δ 46.7 (CH_2), 60.0 (CH_2), 62.2 (CH_2), 66.7 (CH_2), 74.0 (CH), 109.4 (2CH), 119.9 (CH), 120.5 (CH), 120.6 (CH), 121.4 (CH), 130.0 (C), 130.2 (C), 136.6 (C), 136.9 (C), 156.0 (C).
16. The final treatment with $\text{K}_2\text{CO}_3/\text{MeOH}$ is necessary to complete the hydrolysis of the butyrate esters resulting from the cyclization.
17. Spectroscopic data for **1**: ^1H NMR (300 MHz, CDCl_3) δ 3.61–3.68 (m, 1H), 3.73–3.81 (m, 1H), 3.94 (dd, $J = 6.7$, 2.0 Hz, 1H), 4.15 (app t, $J = 8.8$ Hz, 1H), 4.45 (s, 2H), 4.48 (s, 2H), 4.56 (app t, $J = 6.4$ Hz, 1H), 4.95–5.02 (m, 1H), 5.89 (d, $J = 1.7$ Hz, 1H), 6.32–6.33 (m, 2H), 7.04 (dd, $J = 2.6$, 1.7 Hz, 1H), 7.41 (d, $J = 8.8$ Hz, 1H), 7.57 (d, $J = 2.6$ Hz, 1H), 7.68 (dd, $J = 8.8$, 2.6 Hz, 1H), 8.07 (d, $J = 1.7$ Hz, 1H); ^{13}C NMR (75 MHz, CDCl_3) δ 46.9 (CH_2), 48.1 (CH_2), 60.8 (CH_2), 67.3 (CH_2), 71.8 (CH), 96.8 (CH), 110.0 (2CH), 120.1 (CH), 120.8 (2CH), 121.9 (CH), 130.5 (C), 130.9 (C), 136.6 (C), 137.0 (C), 154.8 (C), 158.7 (CH), 163.9 (C).